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[Chemistry Class 11 – \[Bihar & JAC\]](#) Dr. S. C. Rastogi, 2022-07-20 Syllabus : Unit I : Some Basic Concepts of Chemistry, Unit II : Structure of Atom, Unit III : Classification of Elements and Periodicity in Properties, Unit IV : Chemical Bonding and Molecular Structure, Unit V : States of Matter : Gases and Liquids, Unit VI : Chemical Thermodynamics, Unit VII : Equilibrium, Unit VIII : Redox Reactions, Unit IX : Hydrogen, Unit X : s-Block Elements (Alkali and Alkaline earth metals) Group 1 and Group 2 Elements, Unit XI : Some p-Block Elements General Introduction to p-Block Elements, Unit XII : Organic Chemistry–Some Basic Principles and Techniques, Unit XIII : Hydrocarbons Classification of Hydrocarbons, Unit XIV : Environmental Chemistry Content : 1. Some Basic Concepts of Chemistry, 2. Structure of Atom, 3. Classification of Elements and Periodicity in Properties, 4. Chemical Bonding and Molecular Structure, 5. States of Matter, 6. Thermodynamics, 7. Equilibrium, 8. Redox Reactions, 9. Hydrogen, 10. s-Block Elements 11. p-Block Elements, 12. Organic Chemistry–Some Basic Principles and Techniques 13. Hydrocarbons 14. Environmental Chemistry I. Appendix II. Log-antilog Table

[Comprehensive Practical Chemistry XII](#) Dr. N. K. Verma 2011-11-01

[Competition Science Vision 2003-08](#) Competition Science Vision (monthly magazine) is published by Pratiyogita Darpan Group in India and is one of the best Science monthly magazines available for medical entrance examination students in India. Well-qualified professionals of Physics, Chemistry, Zoology and Botany make contributions to this magazine and craft it with focus on providing complete and to-the-point study material for aspiring candidates. The magazine covers General Knowledge, Science and Technology news, interviews of toppers of examinations, study material of Physics, Chemistry, Zoology and Botany with model papers, reasoning test questions, facts, quiz contest, general awareness and mental ability test in every monthly issue.

[Comprehensive Chemistry Activities Vol. I XI](#) Dr. N. K. Verma 2011-12

[Chemistry R P Manchanda A text book on Chemistry](#)

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[ISC Chemistry Book 1 for Class XI \(2021 Edition\)](#) R.D.MADAN 2018 ISC Chemistry Book 1

[ISC Practical Chemistry Volume II for Class XII \(2021 Edition\)](#) Ajaya Baboo 2018 ISC Practical Chemistry Class XII has been thoroughly revised as per the latest syllabus for ISC (Class XII) prescribed by the Council for the Indian School Certificate Examinations (CICSE), New Delhi.

[Comprehensive Practical Physics XI](#) J. N. Jaiswal 2012-08-01

[Conceptual Chemistry Class XI Vol. II](#) S K Jain A book on Conceptual Chemistry

[Practical/Laboratory Manual Chemistry Class – XI](#) Er. Meera Goyal 2021-05-29 1. Basic Laboratory Techniques 1. To cut a glass tube or glass rod, 2. To bend the glass rod at an angle, 3. To draw a glass jet from a glass tube 4. To bore a cork and fit a glass tube into it Viva-Voce 2. Characterisation and Purification of Chemical Substances 1. To determine the melting point of the given unknown organic compound and its identification (simple laboratory technique) Viva-Voce 2. To determine the boiling point of a given liquid when available in small quantity (simple laboratory method) Viva-Voce 3. To prepare crystals of pure potash alum [K₂SO₄.Al₂(SO₄)₃.24H₂O] from the given impure sample 4. To prepare the pure crystals of copper sulphate from the given crude sample 5. To prepare pure crystals of benzoic acid from a given impure sample Viva-Voce 3. Measurement of pH Values 1. To determine the pH value of vegetable juices, fruit juices, tap water and washing soda by using universal pH paper 2. To determine and compare the pH values of solutions of strong acid (HCl) and weak acid (CH₃COOH) of same concentration 3. To study the pH change in the titration of strong base Vs. strong acid by using universal indicator paper 4. To study the pH change by common ion (CH₃COO⁻ ion) in case of weak acid (CH₃COOH) 5. To determine the change in pH value of weak base (NH₄OH) in presence of a common ion (NH₄⁺) Viva-Voce 4. Chemical Equilibrium 1. To study the shift in equilibrium between ferric ions and thiocyanate ions by changing the concentrations of either of the ions 2. To study the shift in equilibrium between [Co(H₂O)₆]²⁺ and Cl⁻ ions by changing the concentrations of either of the ions Viva-Voce 5. Quantitative Analysis 1. To prepare M/10 oxalic acid solution by direct weighing method 2. To prepare M/10 solution of sodium carbonate by direct weighing method 3. To determine the strength of given solution of sodium hydroxide by titrating it against N/10 or M/20 solution of oxalic acid 4. To determine the strength of a given solution of hydrochloric acid by titrating it against a standard N/10 or M/20 sodium carbonate solution Viva-Voce 6. Qualitative Analysis Analysis of Anions Analysis of Cations Viva-Voce 7. Detection of Elements in Organic Compounds 1. To detect the presence of nitrogen, sulphur and halogens in a given organic compound by Lassaigne's test 2. To detect the presence of nitrogen, sulphur and halogens in the given organic compound sample number by Lassaigne's test Viva-Voce INVESTIGATORY PROJECTS 1. Checking of Bacterial Contamination in Water 1. To check the bacterial contamination in drinking water by testing sulphide ions Viva-Voce 2. Methods of Water Purification 1. To purify water from suspended impurities by using sedimentation 2. To purify water by boiling 3. To purify water by distillation method 4. To purify water by reverse osmosis technique 5. To purify water by GAC method 6. To purify water by bleach treatment 7. To purify water by oxidising agent 8. To purify water by ozone treatment method Viva-Voce 3. Water Analysis 1. To test the hardness of different water samples Viva-Voce 4. Foaming Capacity of Various Soaps 1. To compare the foaming capacity of different washing soaps 2. To study the effect of addition of sodium carbonate on foaming capacity of washing soap Viva-Voce 5. Tea Analysis 1. To study the acidity of different samples of tea leaves (tea) by using pH paper Viva-Voce 6. Analysis of Fruits and Vegetable Juices 1. To analyse the fruit and vegetable juices for the constituent present in them Viva-Voce 7. Rate of Evaporation 1. To study the rate of evaporation of different liquids 1 Viva-Voce 8. Effect of Acids and Bases on Tensile Strength of Fibres 1. To compare the tensile strength of natural fibres and synthetic fibres 2. To study the effect of acids and bases on tensile strength of different fibres Viva-Voce

[Appendix to the Journals of the House of Representatives of New Zealand](#) New Zealand. Parliament. House of Representatives 1898

[Practical/Laboratory Manual Chemistry Class XI based on NCERT guidelines by Dr. S. C. Rastogi & Er. Meera Goyal](#) Dr. S. C. Rastogi 2020-06-23 An Excellent Book in Accordance with the latest syllabus for Class-11 Prescribed by CBSE/NCERT and Adopted by Various State Education Boards. (A) Basic Laboratory Techniques – 1. To cut a glass tube or glass rod, 2. To bend the glass rod at an angle, 3. To draw a glass jet from a glass tube, 4. To bore a cork and fit a glass tube into it. (B) Characterisation and Purification of Chemical Substances– 1. To determine the melting point of the given unknown organic compound and its identification (simple laboratory technique), 2. To determine the boiling point of a given liquid when available in small quantity (simple laboratory method), 3. To prepare crystals of pure potash alum [K₂SO₄.Al₂(SO₄)₃.24H₂O] from the given impure sample, 4. To prepare the pure crystals of copper sulphate from the given crude sample, 5. To prepare pure crystals of benzoic acid from a given impure sample. (C) Measurement of pH Values 1. To determine the pH value of vegetable juices, fruit juices, tap water and washing soda by using universal pH paper, 2. To determine and compare the pH values of solutions of strong acid (HCl) and weak acid (CH₃COOH) of same concentration, 3. To study the pH change in the titration of strong base Vs. strong acid by using universal indicator paper, 4. To study the pH change by common ion (CH₃COO⁻ ion) in case of weak acid (CH₃COOH), 5. To determine the change in pH value of weak base (NH₄OH) in presence of a common ion (NH₄⁺), (D) Chemical Equilibrium 1. To study the shift in equilibrium between ferric ions and thiocyanate ions by changing the concentrations of either of the ions, 2. To study the shift in equilibrium between [Co(H₂O)₆]²⁺ and Cl⁻ ions by changing the concentrations of either of the ions, (E) Quantitative Analysis 1. To prepare M/10 oxalic acid solution by direct weighing method, 2. To prepare M/10 solution of sodium carbonate by direct weighing method, 3. To determine the strength of given solution of sodium hydroxide by titrating it against N/10 or M/20 solution of oxalic acid, 4. To determine the strength of a given solution of hydrochloric acid by titrating it against a standard N/10 or M/20 sodium carbonate solution, (F) Qualitative Analysis 1. Analysis of Anions, 2. Analysis of Cations (G) Detection of Elements in Organic Compounds 1. To detect the presence of nitrogen, sulphur and halogens in a given organic compound by Lassaigne's test, 2. To detect the presence of nitrogen, sulphur and halogens in the given organic compound sample number by Lassaigne's test INVESTIGATORY PROJECTS (A) Checking of Bacterial Contamination in Water 1. To check the bacterial contamination in drinking water by testing sulphide ions (B) Methods of Water Purification 1. To purify water from suspended impurities by using sedimentation, 2. To purify water by boiling, 3. To purify water by distillation method, 4. To purify water by reverse osmosis technique. 5. To purify water by GAC method, 6. To purify water by bleach treatment, 7. To purify water by oxidising agent, 8. To purify water by ozone treatment method. (C) Water Analysis 1. To test the hardness of different water samples. (D) Foaming Capacity of Various Soaps 1. To compare the foaming capacity of different washing soaps, 2. To study the effect of addition of sodium carbonate on foaming capacity of washing soap (E) Tea Analysis 1. To study the acidity of different samples of tea leaves (tea) by using pH paper (F) Analysis of Fruits and Vegetable Juices 1. To analyse the fruit and vegetable juices for the constituent present in them (G) Rate of Evaporation 1. To study the rate of evaporation of different liquids (H) Effect of Acids and Bases on Tensile Strength of Fibres 1. To compare the tensile strength of natural fibres and synthetic fibres, 2. To study the effect of acids and bases on tensile strength of different fibres. Log & Antilog Table

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Journal, Appendix New Zealand. Parliament. House of Representatives 1898

Calendar - McGill University McGill University 1904

Comprehensive Practical Chemistry XI Dr. N. K. Verma 2010-02

The Chemical News and Journal of Industrial Science 1873

Guy's Hospital Gazette 1889

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